# Crystallography

- The unit cell
- Space groups
- Bragg's law
- The structure factor

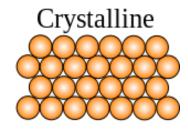
Magnus H. Sørby *Institute for Energy Technology, Norway* 

# **Motivation**

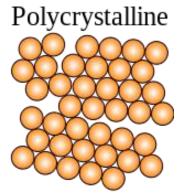
# Single crystals Powder Li<sub>3</sub>AlD<sub>6</sub>, 295 K: PND, PUS

#### Some definitions

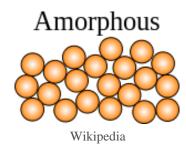
Crystal or crystalline solid: a solid material whose constituents, such as atoms, molecules or ions, are arranged in a highly ordered microscopic structure.



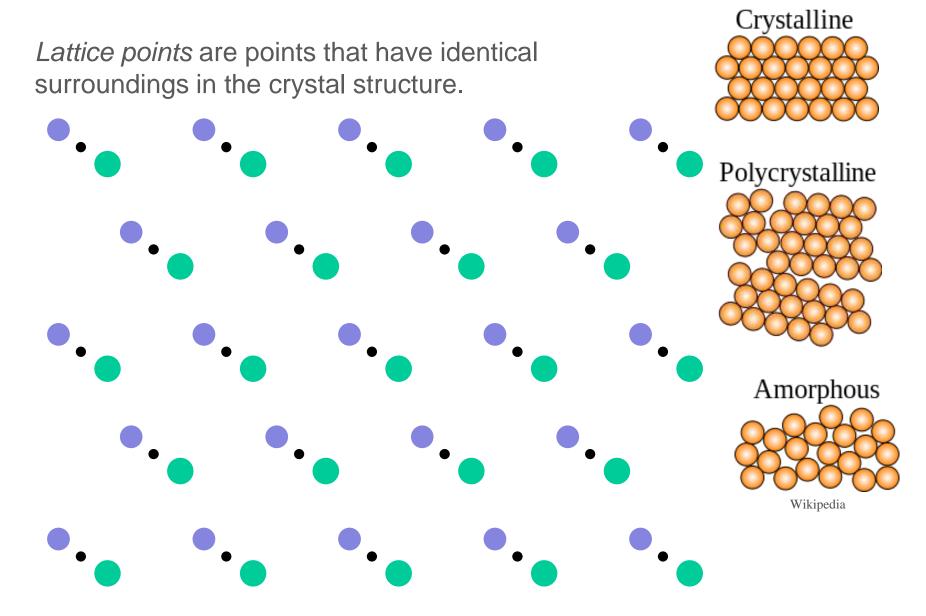
Polycrystalline materials (powder) are solids that are composed of many crystallites of varying size and orientation. Crystallites are also referred to as grains.



Amorphous solids: the atoms have no periodic structure whatsoever.

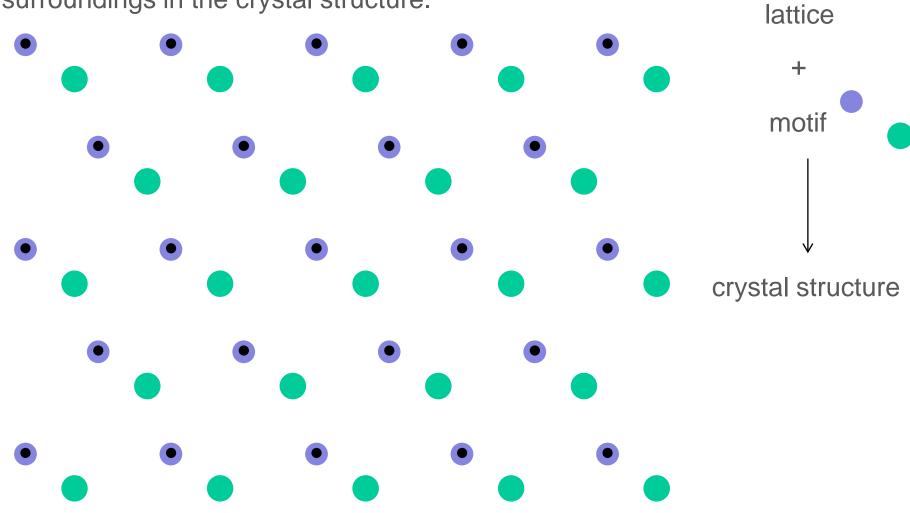


#### **Some definitions**

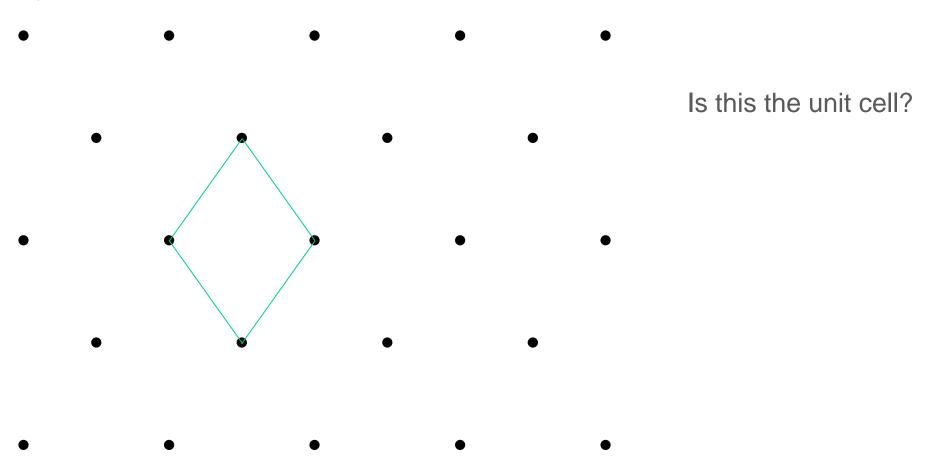


#### **Some definitions**

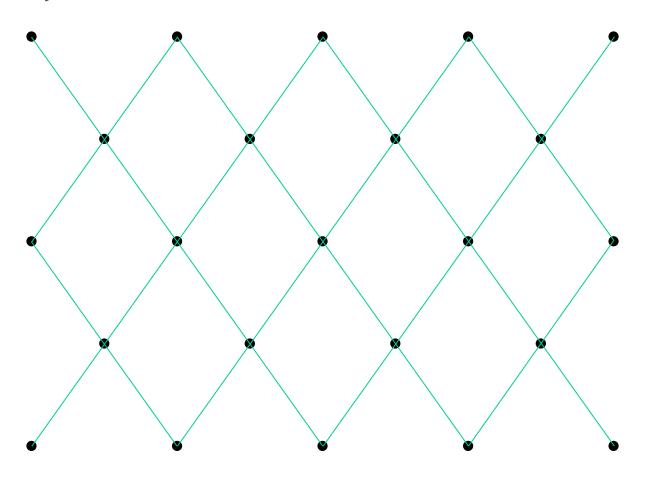
Lattice points are points that have identical surroundings in the crystal structure.



The *smallest* unit ("box") that can be repeated in all directions to build up the crystal structure *and* shows the full symmetry of the crystal structure.

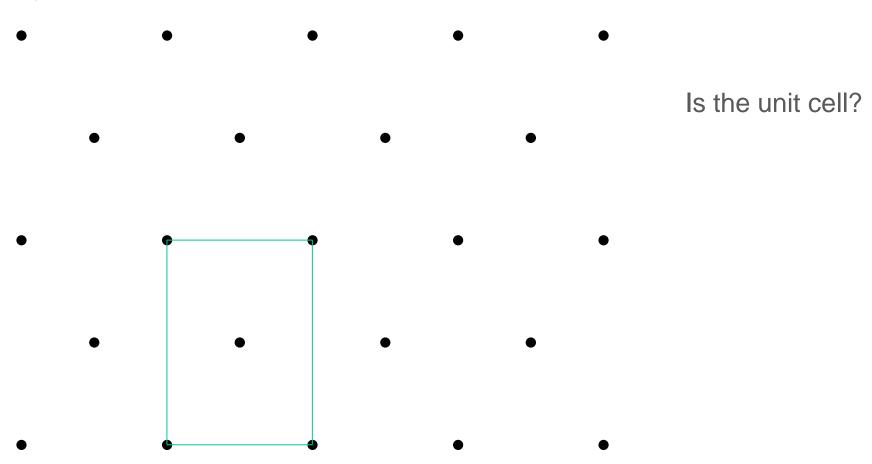


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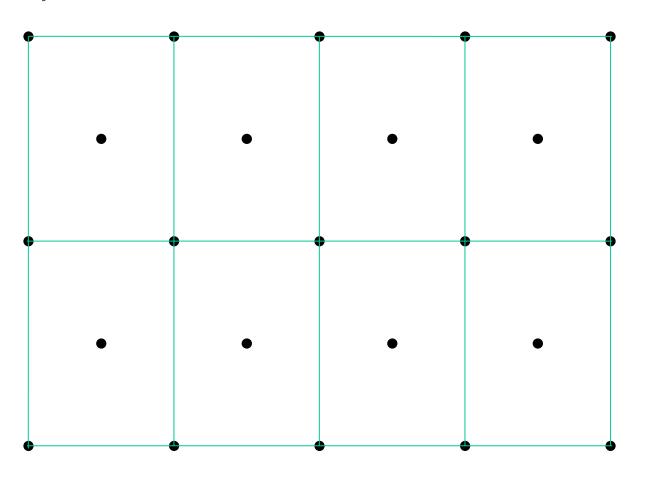


Is this the unit cell?

The *smallest* unit ("box") that can be repeated in all directions to build up the crystal structure *and* shows the full symmetry of the crystal structure.



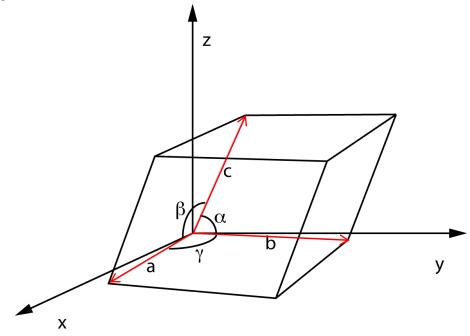
The *smallest* unit ("box") that can be repeated in all directions to build up the crystal structure *and* shows the full symmetry of the crystal structure.



Is the unit cell?

Yes! With «centered rectangular» symmetry

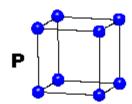
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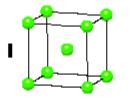


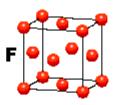
Unit cell dimensions: a, b, c, angles:  $\alpha$ ,  $\beta$ ,  $\gamma$  or defined by three vectors **a**, **b**, **c** 

#### CUBIC

$$a = b = c$$
  
 $\alpha = \beta = \gamma = 90^{\circ}$ 

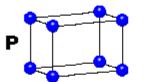


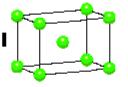




#### **TETRAGONAL**

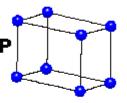
$$a = b \neq c$$
  
 $\alpha = \beta = \gamma = 90^{\circ}$ 

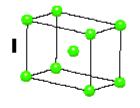


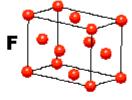


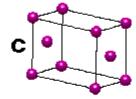
#### **ORTHORHOMBIC**

$$a \neq b \neq c$$
  
 $\alpha = \beta = \gamma = 90^{\circ}$ 



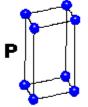


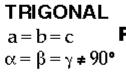


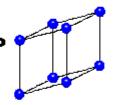


#### **HEXAGONAL**

$$a = b \neq c$$
  
 $\alpha = \beta = 90^{\circ}$   
 $\gamma = 120^{\circ}$ 

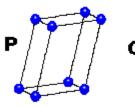


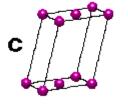




#### MONOCLINIC

$$a \neq b \neq c$$
  
 $\alpha = \gamma = 90^{\circ}$   
 $\beta \neq 120^{\circ}$ 





# Р

#### 4 types of unit cells

- P = Primitive
- = Body-centered
- F = Face-centered
- C = Side-centered

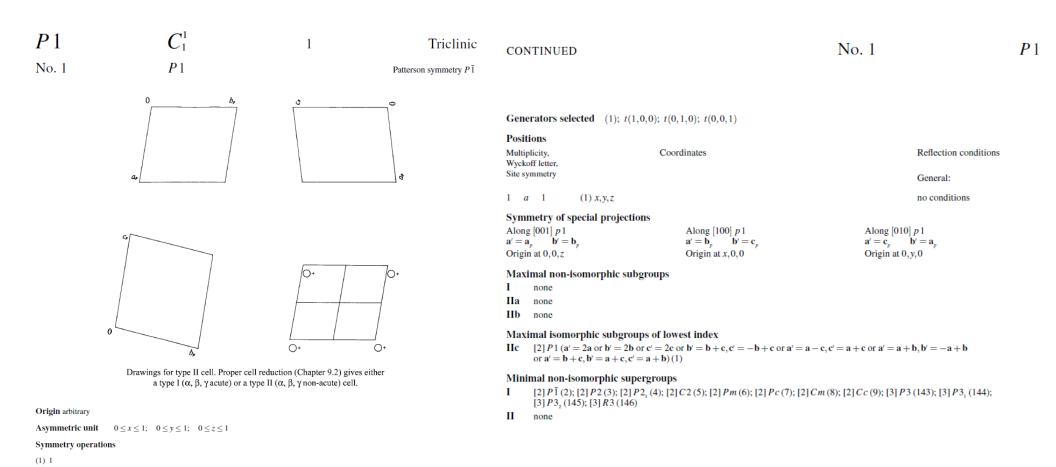
+

- 7 crystal systems
- → 14 Bravais lattices

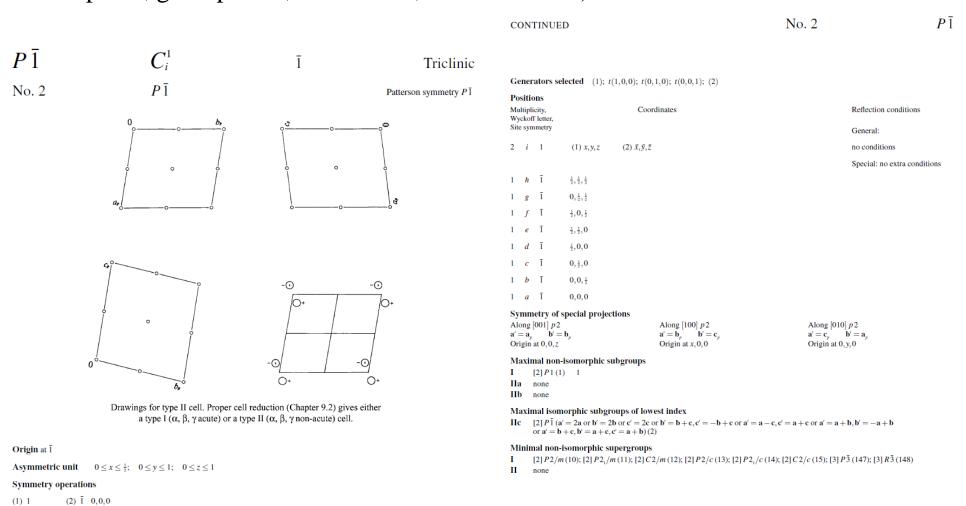
#### TRICLINIC

$$a \neq b \neq c$$
  
 $\alpha \neq \beta \neq \gamma \neq 90^{\circ}$ 

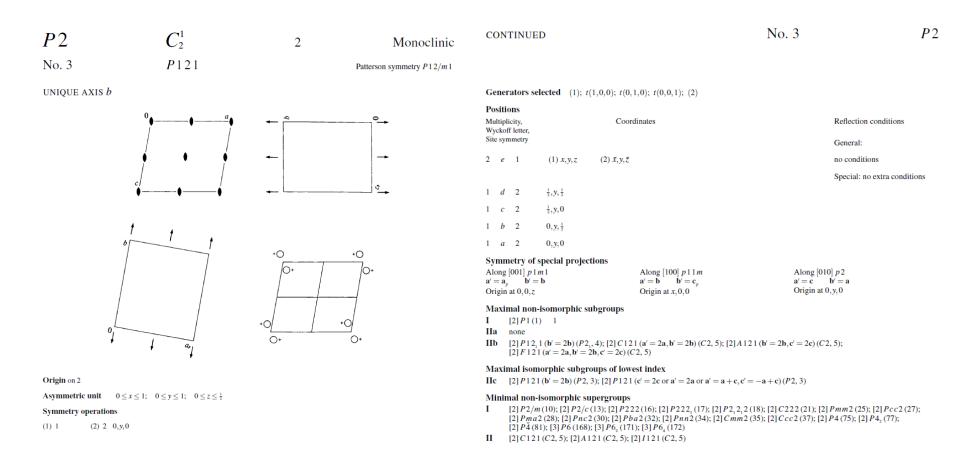
14 Bravais lattices + symmetry elements (centre of symmetry, rotation axes, mirror plane, glide planes, screw axes, inversion axes).



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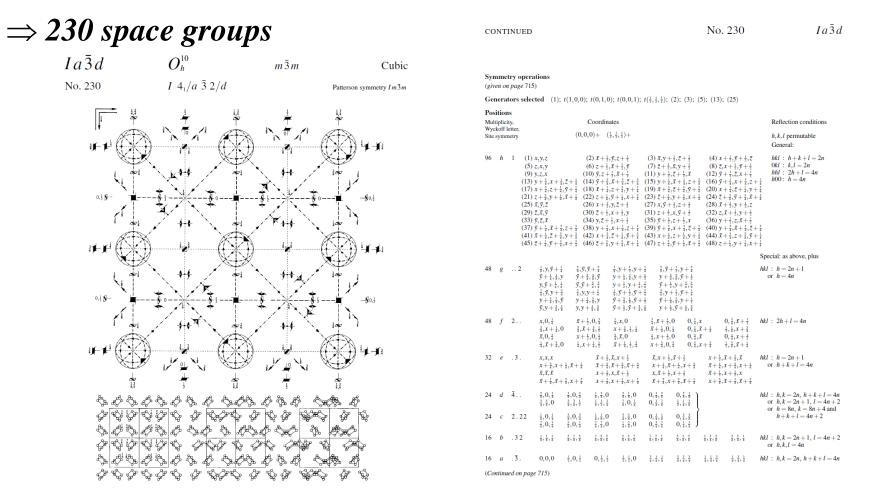


14 Bravais lattices + symmetry elements (centre of symmetry, rotation axes, mirror plane, glide planes, screw axes, inversion axes).



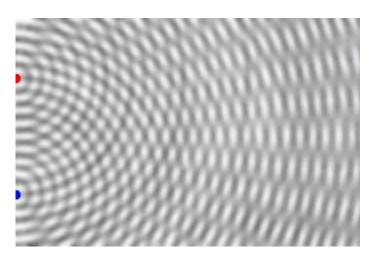
14 Bravais lattices + symmetry elements (centre of symmetry, rotation axes, mirror plane, glide planes, screw axes, inversion axes).

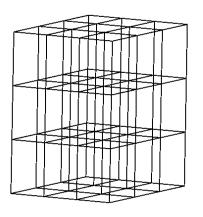
Up to 230 different ways to replicate a finite object in 3-dimensional space.

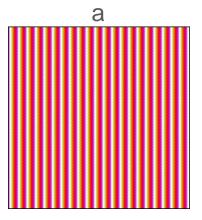


#### **Diffraction**

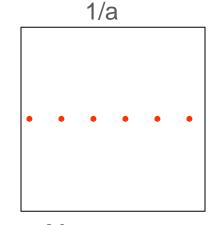
- Laser at slits gives line of diffractions points
  - Repeating units in three dimensions give rise to diffraction pattern
  - Repeating units: Unit cells



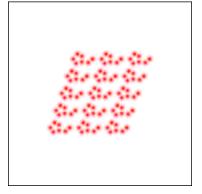


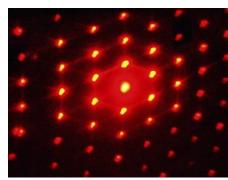


Diffracting object (Real space)

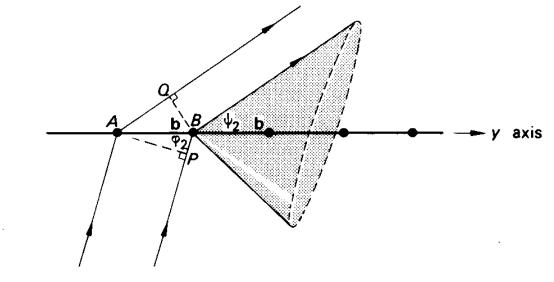


Measurements (Reciprocal space)





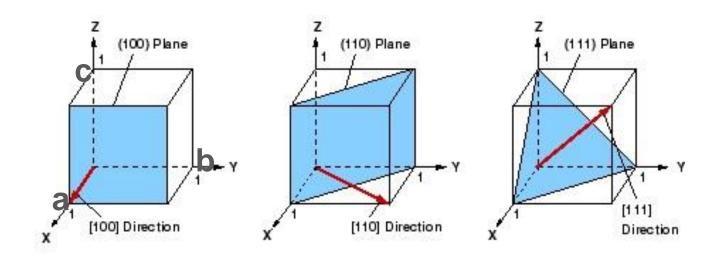
# Direction of diffracted beams: Laue's interference condition



• To get interference (assuming elastic scattering):  $AQ-PB = b(\cos\psi_2 - \cos\phi_2) = m\lambda \quad (m = integer)$ 

• In 3D: 
$$a(\cos\psi_1 - \cos\varphi_1) = h\lambda$$
 
$$b(\cos\psi_2 - \cos\varphi_2) = k\lambda$$
 Laue equations 
$$c(\cos\psi_3 - \cos\varphi_3) = l\lambda$$

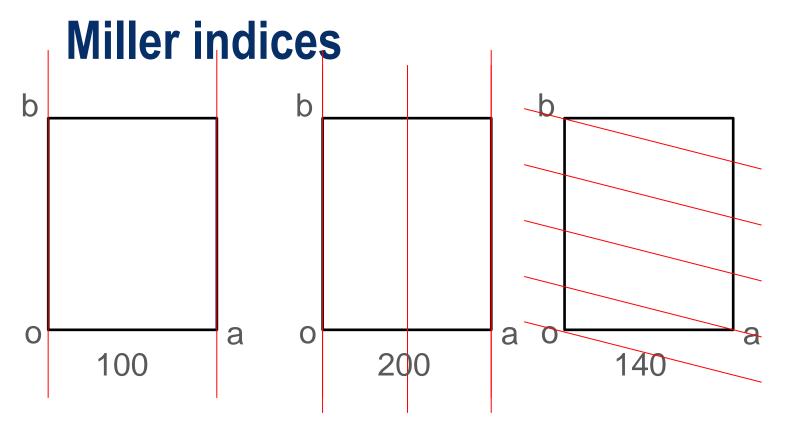
#### Miller indices



The (*hkl*) plane makes intercepts **a**/*h*, **b**/*k* and **c**/*l* along the x, y, z axes.

The direction [hkl] is perpendicular to the (hkl) plane.

a,b,c: Unit cell dimensions

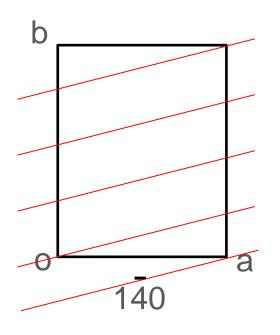


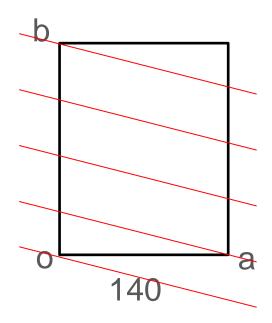
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a,b,c: Unit cell dimensions

#### Miller indices



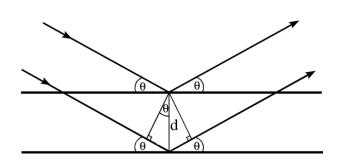


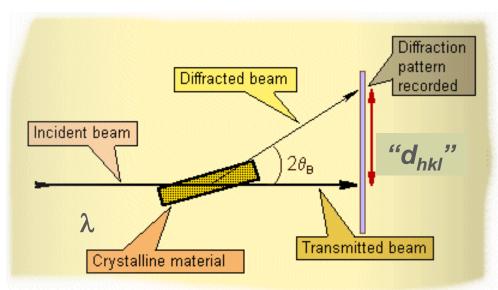
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The direction [hkl] is perpendicular to the (hkl) plane.

a,b,c: Unit cell dimensions

#### Bragg's law





$$2d_{hkl}\sin\theta = \lambda$$

Ex. Orthorhombic crystal system:

$$d_{hkl} = \frac{1}{\sqrt{(\frac{h}{a})^2 + (\frac{k}{b})^2 + (\frac{l}{c})^2}}$$

Cubic: 
$$d_{hkl} = \frac{a}{\sqrt{h^2 + k^2 + l^2}}$$

#### Reciprocal space

#### **Definitions:**

a\*, b\*, c\*: Reciprocal lattice vectors

$$\mathbf{a}^* = \frac{2\pi}{V_c} \ (\mathbf{b} \ x \ \mathbf{c}); \ \mathbf{b}^* = \frac{2\pi}{V_c} \ (\mathbf{c} \ x \ \mathbf{a}); \ \mathbf{c}^* = \frac{2\pi}{V_c} \ (\mathbf{a} \ x \ \mathbf{b})$$

 $V_c$  = volume of the unit cell

a\* perpendicular to the plane containing b and c
b\* perpendicular to plane containing a and c
c\* perpendicular to plane containing a and b

$$\mathbf{a}^* \cdot \mathbf{b} = \mathbf{a}^* \cdot \mathbf{c} = \mathbf{b}^* \cdot \mathbf{a} = \mathbf{b}^* \cdot \mathbf{c} = \mathbf{c}^* \cdot \mathbf{a} = \mathbf{c}^* \cdot \mathbf{b} = 0$$
  
 $\mathbf{a}^* \cdot \mathbf{a} = \mathbf{b}^* \cdot \mathbf{b} = \mathbf{c}^* \cdot \mathbf{c} = 1$ 

#### WHY BOTHER WITH THIS???

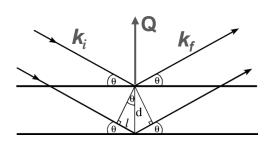
# Real space vs. Reciprocal space

#### Real space

- hkl: Indices for set of planes
- a, b, c: Vectors defining unit cell

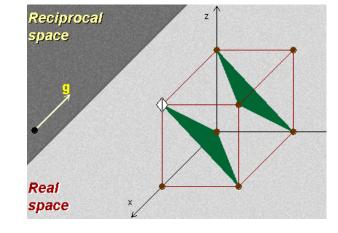
#### Reciprocal space

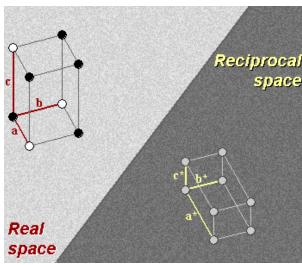
- $H_{hkl} = ha^* + kb^* + c^*$
- hkl: Coordinates of points
- a\*, b\*, c\*: Basal vectors in reciprocal lattice
- $\mathbf{H}_{hkl} \perp (hkl)$
- $|\mathbf{H}_{hkl}| = 2\pi/d_{hkl}$



Elastic scattering:  $|\mathbf{k}_i| = |\mathbf{k}_f| = 2\pi/\lambda$ 

$$\mathbf{Q} \equiv \mathbf{k_f} - \mathbf{k_i} \rightarrow |\mathbf{Q}| = 4\pi \sin(\theta)/\lambda$$
 the scattering vector





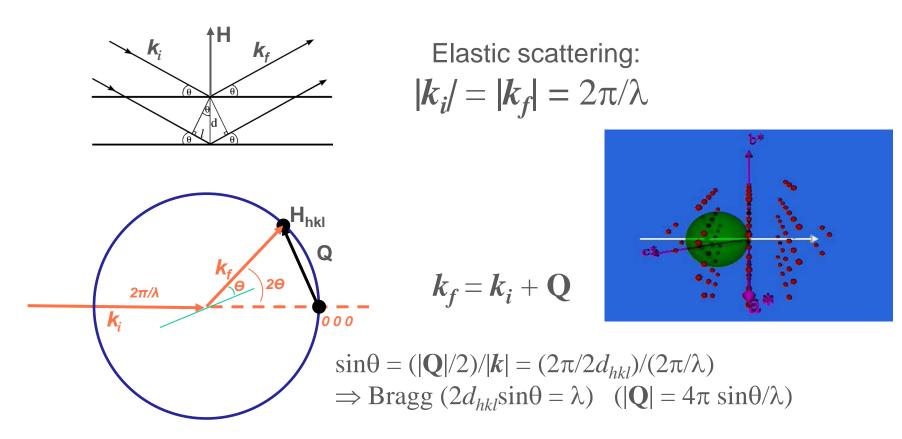
What if 
$$\mathbf{Q} = \mathbf{H}_{hkl}$$
?  
 $4\pi \sin(\theta)/\lambda = 2\pi/d_{hkl}$ 

$$\rightarrow \lambda = 2\pi d_{hkl} \sin(\theta)$$

New condition for constructive interference:  $\mathbf{Q} = \mathbf{H}_{hkl}$ 

# **Ewald's sphere construction**

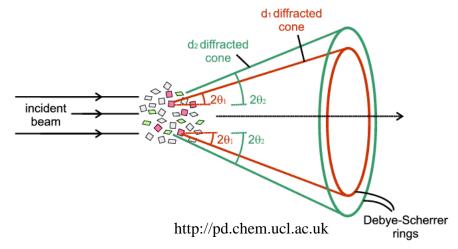
Parallel and monochromatic beam



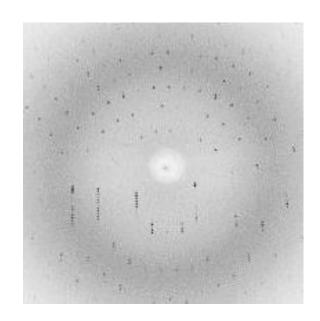
When reciprocal lattice point hkl on Ewald's sphere  $\Rightarrow$  Bragg's law satisfied for the plane (hkl) and Bragg-scattering in that direction.

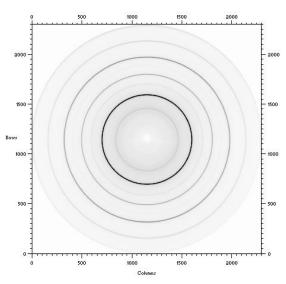
## **Diffraction patterns**

- The reciprocal lattice can be measured
- The positions of the spots
   (for single crystals) or lines
   (for powders) are given from
   Bragg's law

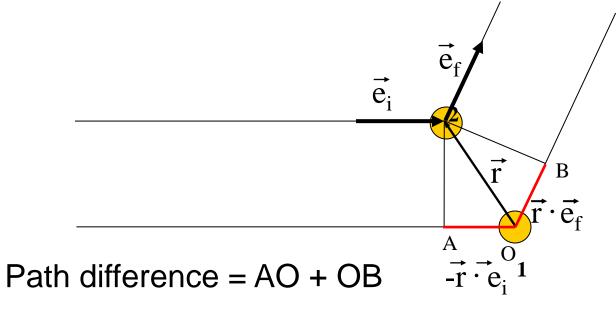


What about the intensities?





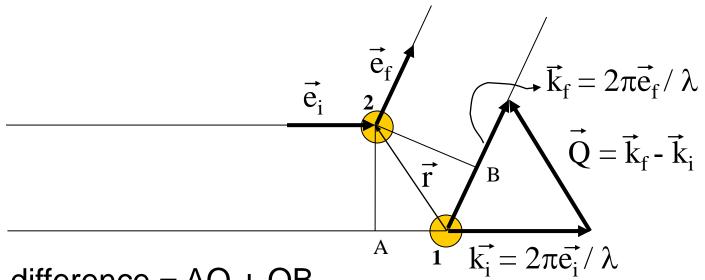
# **Scattering**



$$=(\vec{e}_f - \vec{e}_i)\cdot\vec{r}$$

Phase difference = 
$$\frac{2\pi(\vec{e}_f - \vec{e}_i)\cdot\vec{r}}{\lambda}$$

### **Scattering**



Path difference = AO + OB

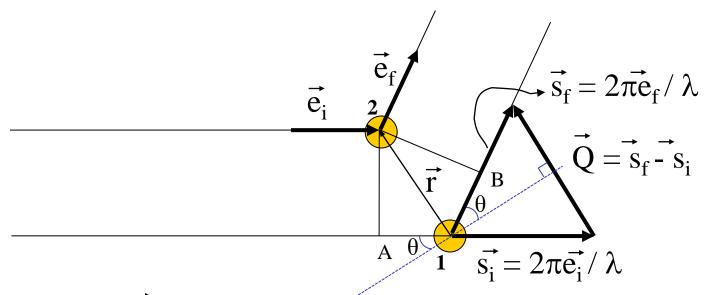
$$= (\vec{e}_f - \vec{e}_i) \cdot \vec{r}$$

Phase difference = 
$$\frac{2\pi(\vec{e}_f - \vec{e}_i)\cdot\vec{r}}{\lambda}$$

$$= \vec{Q}\vec{\tau}$$

amplitude phase
$$F(\vec{Q}) = \sum_{j}^{N} b_{j} e^{i\vec{Q}\cdot\vec{r}_{j}}$$

### **Scattering**



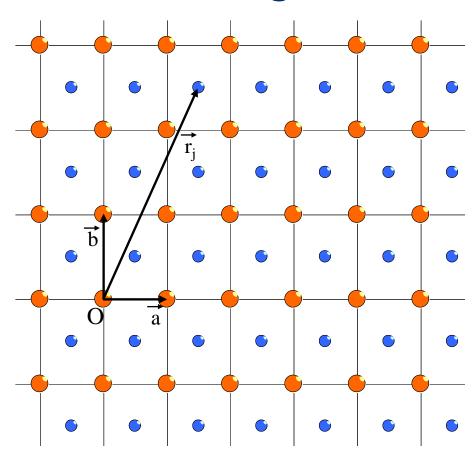
#### Properties of Q

Q is perpendicular to the scattering plane

$$- |\mathbf{Q}| = \frac{4\pi \sin(\theta)}{\lambda} = \frac{2\pi}{d}$$

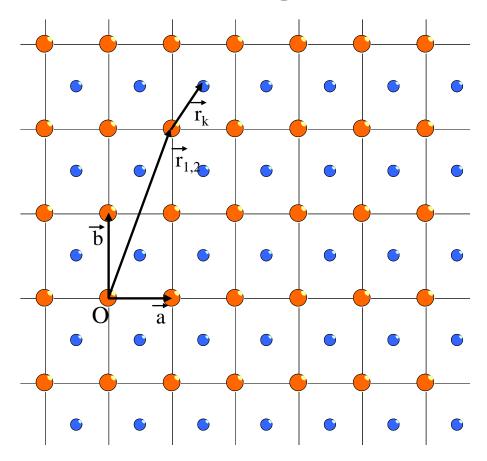
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# Scattering from ideal crystals



$$F(\vec{Q}) = \sum_{i}^{N} b_{i} e^{i\vec{Q}\cdot\vec{r}_{i}}$$

# Scattering from ideal crystals



$$F(\vec{Q}) = \sum_{j}^{N} b_{j} e^{i\vec{Q}\cdot\vec{r}_{j}}$$

$$= \sum_{u,v,w=-\infty}^{\infty} \sum_{k}^{n} b_{k} e^{i\vec{Q}\cdot(\vec{r}_{k} + \vec{r}_{uvw})}$$

$$= \sum_{k=1}^{n} b_{k} e^{i\vec{Q}\cdot\vec{r}_{k}} \sum_{u,v,w=-\infty}^{\infty} e^{i\vec{Q}\cdot\vec{r}_{uvw}}$$

$$=\sum_{k=1}^{n}b_{k}\ e^{i\vec{Q}\cdot\vec{r_{k}}}\cdot\frac{1}{V}\sum_{h,k,l=-\infty}^{\infty}\delta(\vec{Q}-\vec{Q}_{hkl})$$

The reciprocal lattice

#### The structure factor

$$F(\mathsf{Q}) = \sum_{j} b_{j} \cdot e^{\mathrm{i}\vec{\mathsf{Q}}\cdot\vec{\mathsf{r}}_{j}}$$

Neutrons: 
$$b_j$$
 = scattering length

X-rays: 
$$b_j = f_{at,j} = \text{atomic form factor}$$

$$F_{hkl} = \sum_{\text{Unit}} b_{\mathbf{j}} \cdot e^{2\pi \mathbf{i}(h \cdot \mathbf{x}_{\mathbf{j}} + k \cdot \mathbf{y}_{\mathbf{j}} + l \cdot \mathbf{z}_{\mathbf{j}})}$$

Intensity of scattered beams:

$$I \sim |F|^2$$

Displacement factor (Debye Waller factor, temperature factor):

$$T_{j} = e^{-W_{j}}, W_{j} = 8\pi^{2} \langle u_{j}^{2} \rangle \frac{\sin^{2} \theta}{\lambda^{2}} = 1/2 Q^{2} \langle u_{j}^{2} \rangle$$

In total:

$$F_{hkl} = \sum_{\substack{\text{Unit} \\ \text{cell}}} b_{j} \cdot e^{2\pi i(h \cdot x_{j} + k \cdot y_{j} + l \cdot z_{j})} \cdot e^{-1/2Q^{2} \langle u_{j}^{2} \rangle}$$

# Structure factor calculation (joint effort ©)

Task: Calculate the expression for the structure factor of body-centered cubic Fe.

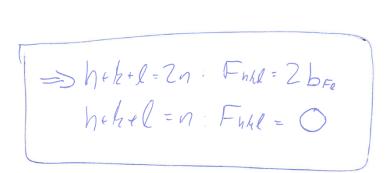
$$F_{hkl} = \sum_{j} b_{j} e^{2\pi i \left(h \cdot x_{j} + k \cdot y_{j} + l \cdot z_{j}\right)}$$

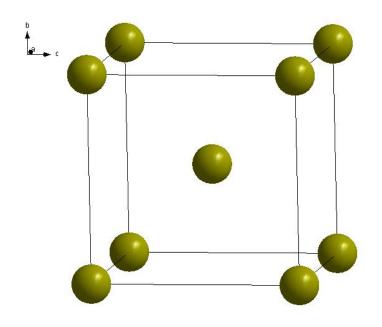
$$= b_{Fe} e^{2\pi i \left(h \cdot o + k \cdot o + l \cdot o\right)} + b_{Fe} e^{2\pi i \left(h \cdot x_{j} + k \cdot x_{j} + l \frac{w}{h}\right)}$$

$$= b_{Fe} \left[1 + e^{i \cdot x_{j} + k \cdot l}\right]$$

$$= 1 + h + h + l = 2n(wen)$$

$$= -l + h + h + l = n(wen)$$





Fe in 0 0 0 and ½ ½ ½ ½

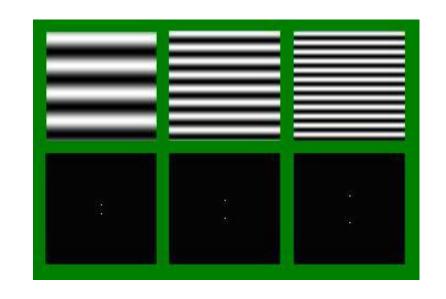
# The phase problem

Definition

$$FT[f(x)] = \widetilde{f}(k) = \int f(x)e^{ikx}dx$$

Inverse Fourier transform:

$$FT^{-1}[\widetilde{f}(k)] = \int \widetilde{f}(k)e^{-ikx}dk$$



The structure factor is the Fourier transformation of:

- The density of atom cores (scattering lengths) for neutrons
- The electron density for X-rays

#### The phase problem

$$F_{hkl} = |F_{hkl}| e^{i\varphi_{hkl}}$$

$$I_{hkl} \propto |F_{hkl}|^2$$

• The nuclear density:

$$\rho(x, y, z) = \tilde{F}_{hkl} = \frac{1}{V_c} \sum_{h} \sum_{k=-\infty}^{\infty} \sum_{l} |F_{hkl}| e^{i\varphi_{hkl}} e^{-2\pi i(hx + ky + lz)}$$

• It is impossible to calculate  $\rho(x,y,z)$  as inverse Fourier transformation of the structure factors as long as  $\phi_{hkl}$  are unknown

⇒ The phase problem in crystallography

## Methods to solve the phase problem

• **Patterson function**: Fourier transformation of  $|F_{hkl}|^2$  without phases:

$$P(u, v, w) = \frac{2}{V_c} \sum_{h} \sum_{k=-\infty}^{\infty} \sum_{k} |F_{hkl}|^2 \cos 2\pi (hu + kv + lw)$$

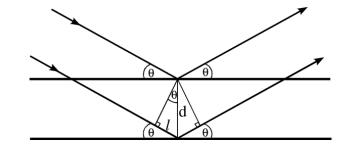
Self-convolution. Peaks in the Patterson function represent a vector between two atoms; weighted by the product of their scattering power. Used to determine positions of strong scatterers.

 Direct methods: Uses statistic relations between observed intensities to guess on phases. Very efficient when scatterers are of similar strength and many independet structure factor amplitues are collected (single crystal)

# Most important points

 The condition for constructive interference (Bragg scattering) is given by Bragg's law

$$\lambda = 2d_{hkl}\sin\theta$$



or the condition that the scattering vector  $\mathbf{Q}$  equals a resiprocal lattice vector:  $\mathbf{Q} = \mathbf{Q}_{hkl}$  (Laue condition)

#### **Most important points**

The scattered wave from a collection of atoms is descibed by the structure factor

$$F(\vec{Q}) = \sum_{j}^{N} b_{j} e^{i\vec{Q}\cdot\vec{r}_{j}}$$

For atoms in an infinitely repeating lattice, F(Q) is non-zero only when  $Q = Q_{hkl}$ 

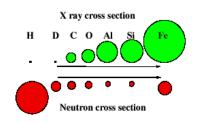
$$F(\vec{Q}_{hkl}) = \sum_{j}^{N} b_j e^{2\pi i(hx_j + ky_j + lz_j)}$$

And of course:  $I \propto |\mathbf{F}|^2$ 

# Power Neutron Diffraction vs Powder X-ray Diffraction

# The glory of neutrons

- There is no systematic correlation between atomic number and the scattering length.
  - Can get information about light and heavy elements simultaneously.

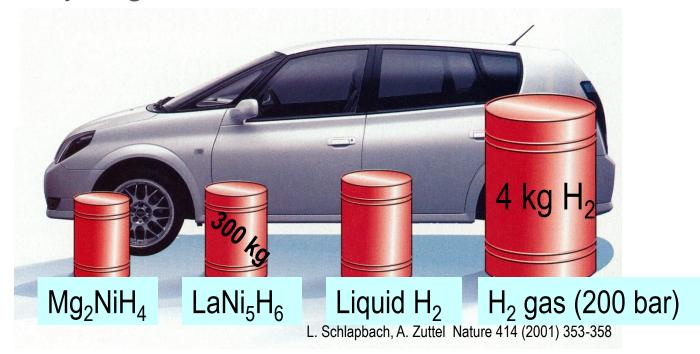


The neutron interacts weakly with matter.

The neutron has a magnetic moment.

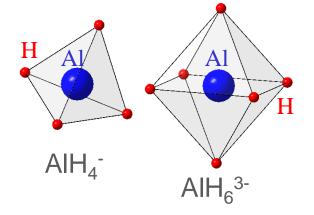
#### **Metal hydrides**

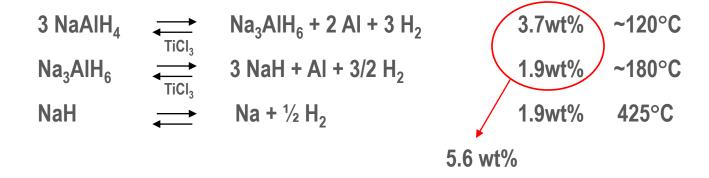
Materials that contain chemical bonding between metal- and hydrogen atoms.



$$M_{(s)} + x/2 H_{2(g)} \leftrightarrow MH_{x(s)} + energy$$

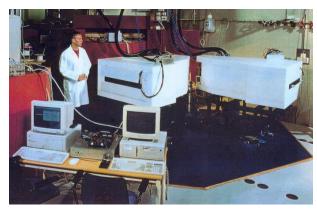
#### **Alanates**





#### **Crystal structure of alanates**

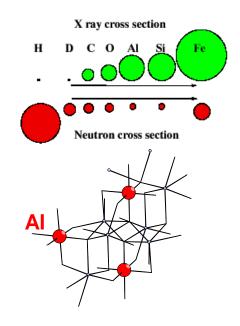
NaAlH<sub>4</sub> Na<sub>3</sub>AlH<sub>6</sub> LiAlH<sub>4</sub> β-LiAlH<sub>4</sub> Li<sub>3</sub>AIH<sub>6</sub> KAIH<sub>4</sub>  $Mg(AIH_4)_2$ Sr<sub>2</sub>AlH<sub>7</sub> BaAlH<sub>5</sub> Ba<sub>2</sub>AlH<sub>7</sub> Na<sub>2</sub>LiAlH<sub>6</sub> K<sub>2</sub>NaAlH<sub>6</sub>  $LiMg(AIH_4)_2$ LiMgAIH<sub>6</sub>  $Ca(AID_4)_2$ CaAID<sub>5</sub>



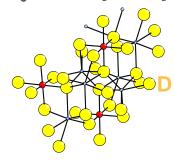
**PUS - high resolution diffractometer** 



The JEEPII reactor



Li<sub>3</sub>AID<sub>6</sub> seen by X-rays



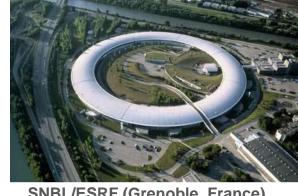
Li<sub>3</sub>AID<sub>6</sub> seen by neutrons

#### **Crystal structure of alanates**

NaAlH<sub>4</sub> Na<sub>3</sub>AlH<sub>6</sub> **LiAIH**<sub>₄</sub> β-LiAlH₄ Li<sub>3</sub>AIH<sub>6</sub> KAIH<sub>4</sub>  $Mg(AIH_4)_2$ Sr<sub>2</sub>AlH<sub>7</sub> BaAlH<sub>5</sub> Ba<sub>2</sub>AlH<sub>7</sub> Na<sub>2</sub>LiAlH<sub>6</sub> K<sub>2</sub>NaAlH<sub>6</sub>  $LiMg(AIH_4)_2$ LiMgAIH<sub>6</sub>  $Ca(AID_4)_2$ CaAID<sub>5</sub>



PUS - high resolution diffractometer

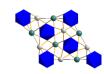


**SNBL/ESRF** (Grenoble, France)













LiAIH<sub>4</sub>

Na<sub>2</sub>LiAlH<sub>6</sub>

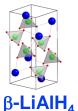
Li<sub>3</sub>AIH<sub>6</sub>

LiMgAIH<sub>6</sub>

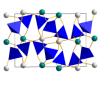
K<sub>2</sub>NaAlH<sub>6</sub>

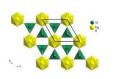
 $Ca(AIH_4)_2$ 

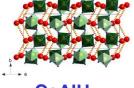












**KAIH**₄

 $LiMg(AIH_4)_3$ 

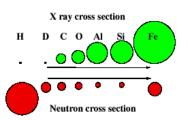
 $Mg(AIH_4)_2$ 

CaAlH<sub>5</sub>

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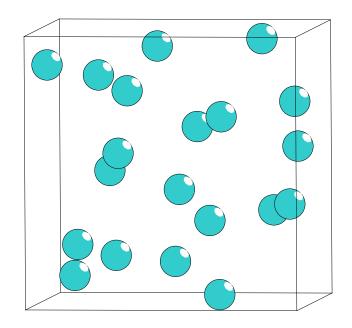
- There is no systematic correlation between atomic number and the scattering length.
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β-Mn: Cubic, complex structure, a = 6.31 Å, Z = 20

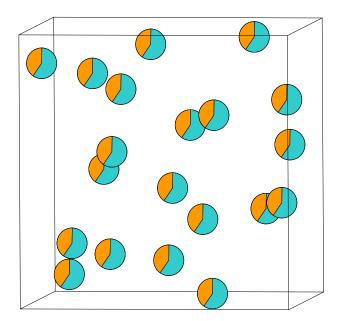
 $Mn(1)_{12}Mn(2)_{8}$ 



What happens when 40% of the Mn is substituted with Co?

β-Mn: Cubic, complex structure, a = 6.31 Å, Z = 20

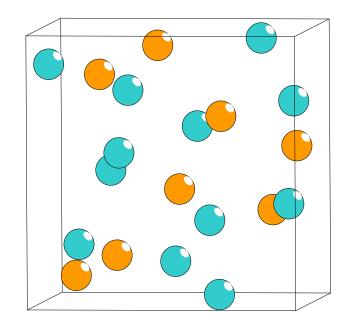
 $[Mn_{0.6}Co_{0.4}](1)_{12}[Mn_{0.6}Co_{0.4}](2)_{8}$ 



What happens if 40% of the Mn is substituted with Co?

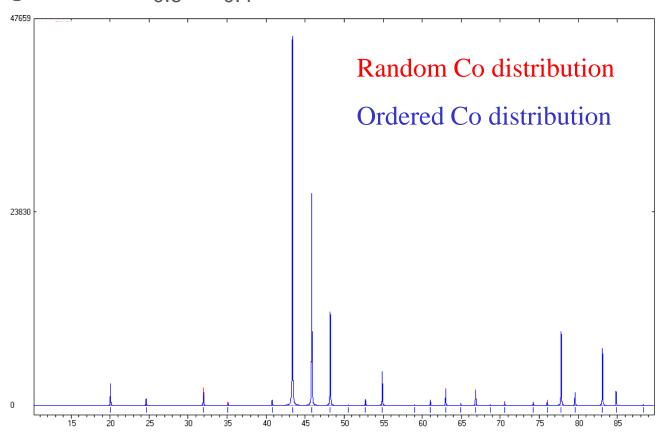
β-Mn: Cubic, complex structure, a = 6.31 Å, Z = 20

 $Mn(1)_{12}Co(2)_8$ 



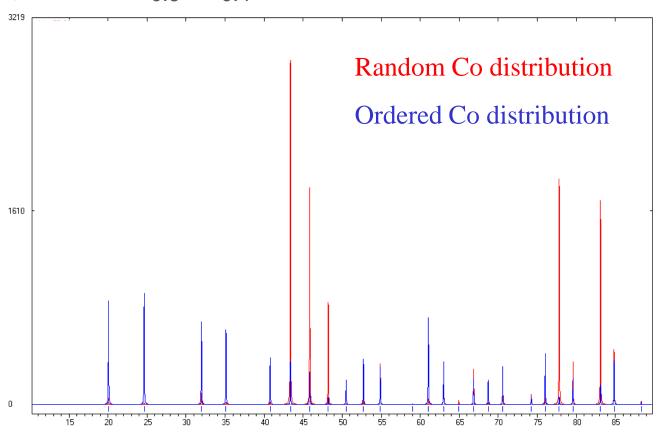
What happens if 40% of the Mn is substituted with Co?

Which model is right for Mn<sub>0.6</sub>Co<sub>0.4</sub>?



X-rays: Z(Mn)=25 Z(Co)=27

Which model is right for Mn<sub>0.6</sub>Co<sub>0.4</sub>?



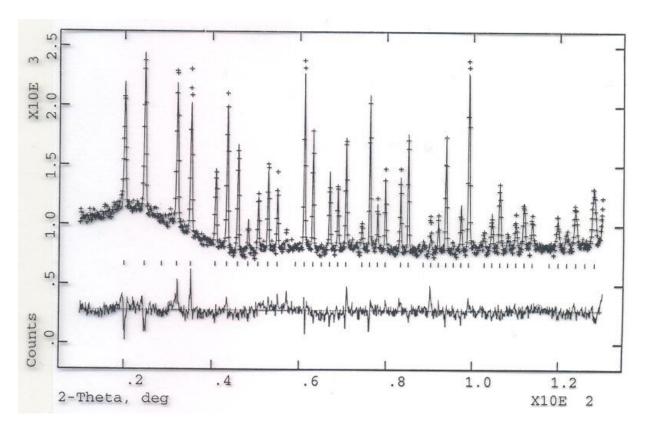
X-rays: Z(Mn)=25 Z(Co)=27

Neutrons:

b(Mn) = -0.373

b(Co) = +0.249

Which model is right for Mn<sub>0.6</sub>Co<sub>0.4</sub>?



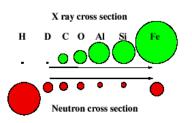
Co selectively occupy the 8-fold position!

O. B. Karlsen, et al. J. Alloys Comp., 2009, 476 (2009) 9-13

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#### **Penetration**

5 mm Al

X-ray 
$$\lambda = 1.54 \text{ Å}$$

$$I/I_0 = 10^{-29}$$

Neutrons 
$$\lambda = 1.0 \text{ Å}$$

$$I/I_0 = 0.996$$

X-ray 
$$\lambda = 0.2 \text{ Å}$$

$$I/I_0 = 0.02$$

#### Sample environment

 Neutrons can penetrate several millimeters of materials like aluminium and steel.



Sample container (Inconel super-alloy) rated to 3000 bar and 600°C.

#### Sample environment

Neutrons can penetrate several millimeters of materials like aluminium and steel.



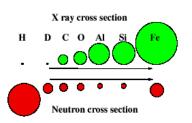
**Furnace** 



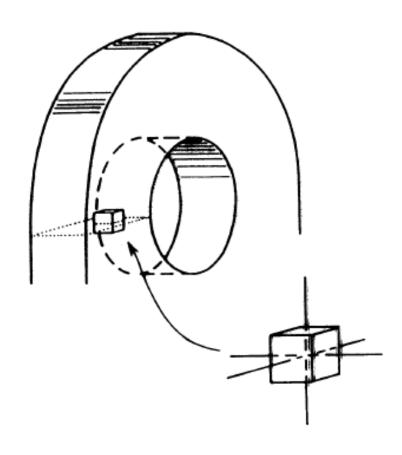
Cryostat

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  - Complicated sample environment is possible.
  - Large samples can be studied.
- The neutron has a magnetic moment.

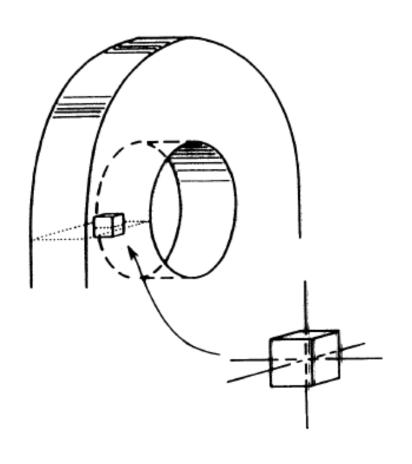


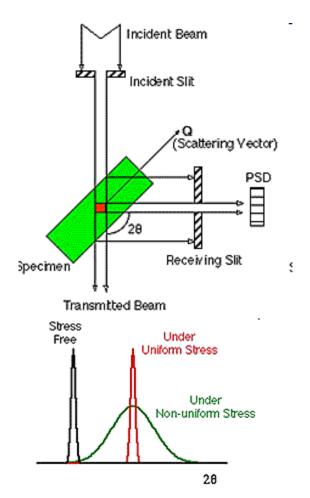
# Study of large samples



 How did the machining of the hole influence the material?

# Study of large samples





#### Study of large samples

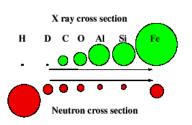


3D residual stress-field can be mapped in a non-destructive way!

Loading a sample at the NRSF2 instrument at Oak Ridge National Lab (US)

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  - Large samples can be studied.
  - Easy interpreations of scattering intensities.
- The neutron has a magnetic moment.



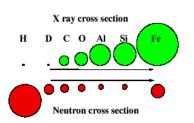
#### **Scattering intensity**

$$I = \left| F_K \right|^2 = \left| \sum_i b_i \cdot e^{i(\vec{r}_i \cdot \vec{Q})} \right|^2 = \left| \sum_i b_j \cdot e^{2\pi i(hx_j + ky_j + lz_j)} \right|^2$$

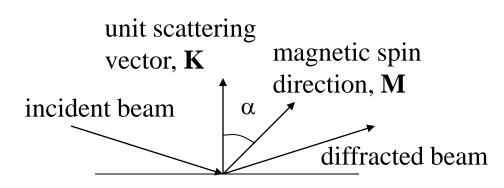
 Can (usually) neglect effects of multiple scattering and absorption.

#### The glory of neutrons

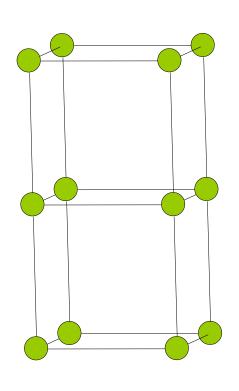
- There is no systematic correlation between atomic number and the scattering length.
  - Can get information about light and heavy elements simultaneously.
  - Can distinguish neighboring elements in the periodic table.
- The neutron interacts weakly with matter.
  - Complicated sample environment is possible.
  - Large samples can be studied.
  - Easy interpreations of scattering intensities.
- The neutron has a magnetic moment.
  - Can study magnetic ordering.

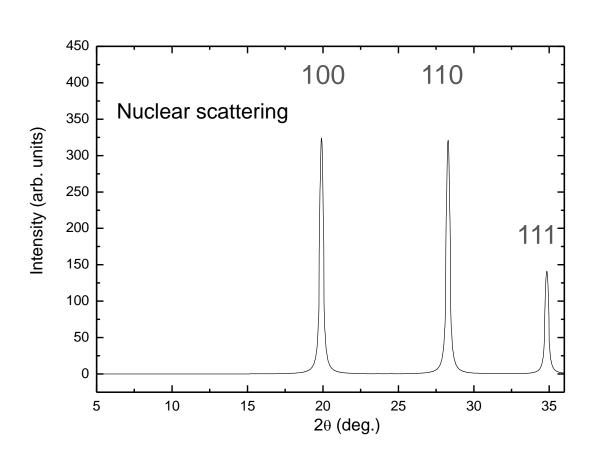


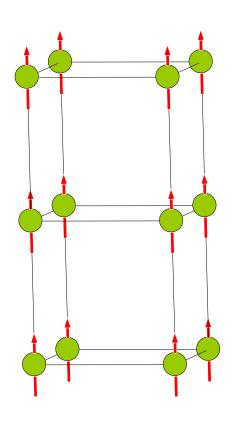
- The neutron has a magnetic moment.
- This will interact with the magnetic moment of atoms with unpaired electrons.

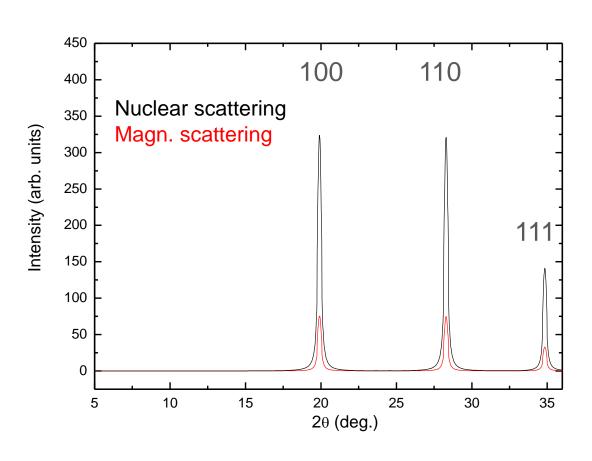


$$\begin{split} \vec{F}_{magnetichkl} &= \sum_{i} \vec{m}_{i} f_{i} \cdot e^{2\pi i (hx_{i} + ky_{i} + lz_{i})} \\ \vec{m} &= \vec{K} (\vec{K} \cdot \vec{M}) - \vec{M}, \qquad |\vec{m}| = \sin \alpha \\ |\vec{m}| &= 0, \qquad \vec{K} || \vec{M} \\ |\vec{m}| &= 1, \qquad \vec{K} \perp \vec{M} \qquad I_{hkl} \propto |F_{hkl}|^{2} = |F_{nucl,hkl}|^{2} + |F_{magnetichkl}|^{2} \end{split}$$

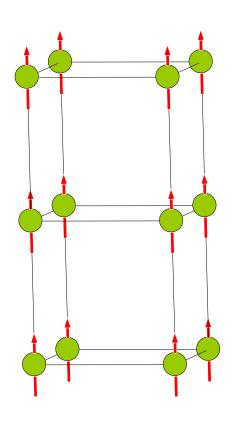


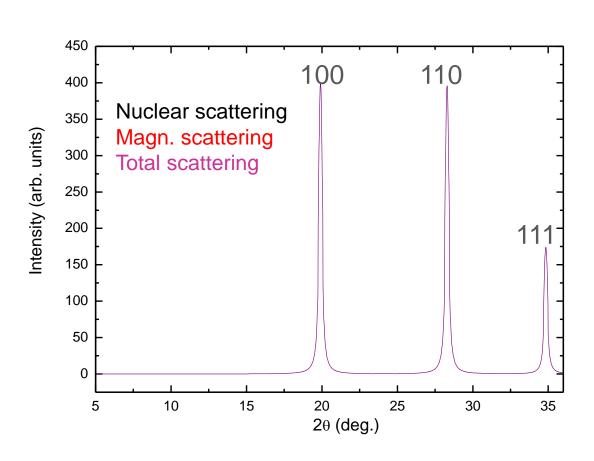




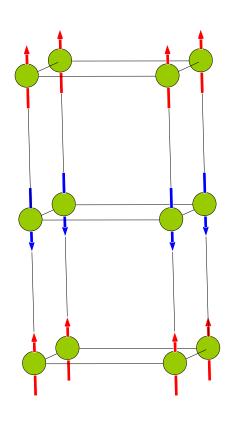


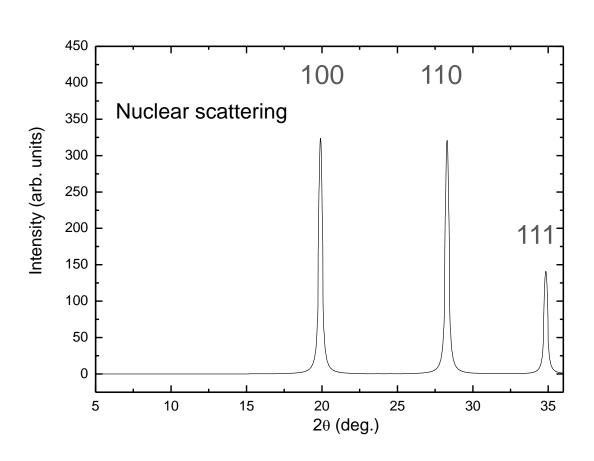
Ferromagnetic



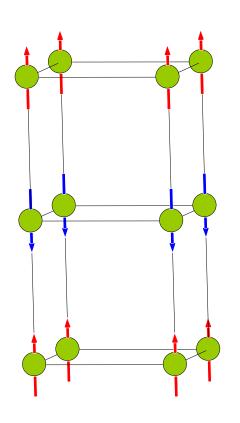


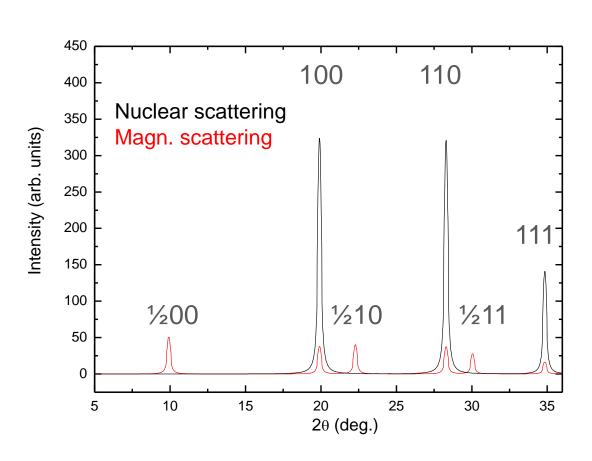
Ferromagnetic



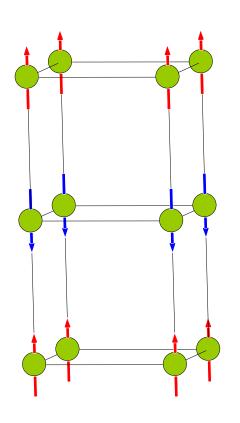


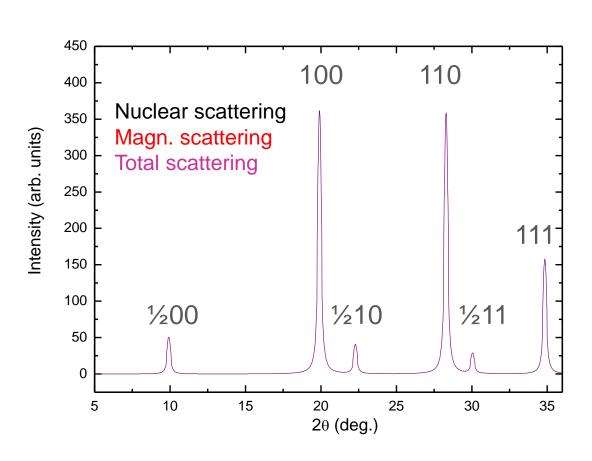
Antiferromagnetic





Antiferromagnetic





Antiferromagnetic

#### Conclusion

Powder neutron diffraction has some major differences from powder X-ray diffraction:

- Different contrast
  - Information about light and heavy elements at the same time.
  - Often good contrast from elements with similar atomic number.
- Weak interactions
  - Scattering from bulk
  - Easy interpreation of intensities
- Magnetic scattering